SPE **PRODUCTS CATALOG**

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INTRODUCTION OF SPETECHNOLOGY

Overview

Solid Phase Extraction (SPE) is a widely used and increasingly popular sample preparation technique. It is mostly used to process liquid samples, extracting, concentrating, and purifying semi-volatile and non-volatile compounds, though it can also be applied to solid samples, which need to be converted into liquids first. At present, SPE is primarily used in the field of food safety, such as for the analysis of antibiotic and antimicrobial residues in various food products, pesticide residue analysis in agricultural products, and the detection of legal and illegal additives in foods. In the pharmaceutical research field, SPE is widely applied in drug metabolism and pharmacokinetics studies, as well as in the analysis of traditional Chinese medicine. In environmental protection, it is used to analyze polycyclic aromatic hydrocarbons (PAHs), polychlorinated biphenyls (PCBs), various pesticides, and organic substances in drinking water, groundwater, and wastewater.

Principle of SPE

Solid Phase Extraction (SPE) is a sample preparation technique that started developing in the mid-1980s. It evolved from liquidsolid extraction and liquid chromatography technologies. The main principle is based on the selective adsorption and desorption of sample components by solid-phase packing materials, achieving separation, purification, and enrichment of the sample. The primary goal is to reduce matrix interference and improve detection sensitivity. When complex sample solutions pass through the adsorbent, the adsorbent selectively retains target compounds and a small number of interfering substances with similar properties through polar interactions, non-polar interactions, or ion exchange. Other components flow through the column. The target compound is then eluted with a stronger solvent system, achieving separation, purification, and enrichment of complex samples. The mechanism of SPE is the same as liquid chromatography; however, due to the special selection of the solvent system in SPE, there are some differences in separation efficiency between the two methods.

1. Advantages of SPE Compared to Traditional Liquid-liquid Extraction

- Higher sample throughput: SPE allows for processing more samples within the same amount of time.
- Better purification: SPE provides effective purification without emulsification issues.
- Enrichment of analytes: It enables the concentration of analytes, improving recovery rates and reproducibility.
- Solvent savings: SPE uses less solvent, making it safer for both the environment and operators.
- Easy automation: SPE is easier to automate compared to traditional LLE.



Liquid-liquid Extration

2. Effect of SPE

Specifically, technology serves the following three purposes Enrichment: In trace analysis or preparation, enriching the target compounds is a necessary process. For example, when analyzing PAHs in water, 1000 mL of a water sample can be passed through an SPE column, where the PAHs are retained, and then eluted with a small amount of solvent (such as 2 mL). This results in a 500-fold concentration of PAHs, meaning that under the same detection conditions, the method detection limit of the analyte is only 1/500 of what it was before treatment.

Purification: This step removes interfering substances prior to instrumental analysis. On one hand, it prevents interference with the target compounds, improving analytical sensitivity; on the other hand, it protects the instrument from damage caused by impurities.

Solvents for conversion: Solvent Exchange: Some analytical instruments have specific requirements for the sample solution's solvent. SPE columns can be used for solvent exchange. For example, when analyzing semi-volatile pollutants in water by GC, directly injecting water can affect separation and damage the gas chromatography column. Therefore, a solvent exchange is needed. The water sample is passed through a reverse-phase extraction column, where the target compounds are retained and separated from the water. The compounds are then eluted with a volatile organic solvent that dissolves the target compounds well. After drying and concentrating, the sample can be analyzed.

3. Composition of SPE

Common SPE columns consist of three parts: the column tube, frits, and adsorbent. Column tube: This is the support for the adsorbent, typically made of polypropylene and usually shaped like a syringe. Yicozoo also provides glass column tubes for special analyses (such as PAES analysis). The outlet size at the bottom of the column is standardized to be compatible with SPE manifolds and instruments from various manufacturers.

Frit: These fix the adsorbent in place and filter the solution. Polyethylene is the most common material for frits, but for special analyses, materials such as PTFE, stainless steel, or glass can also be used.

Adsorbent: The core component that performs the separation in SPE columns. Adsorbents like Florisil and graphitized carbon black, which were widely used in early column chromatography, are still employed in SPE. Currently, the most common adsorbents are silica-bonded ones, made by bonding various functional groups to silica particles. In recent years, organic polymer adsorbents, such as polydivinylbenzene-N-vinylpyrrolidone, have been developed. These adsorbents are increasingly replacing silica-bonded ones in many applications due to their excellent reproducibility, wide pH range, and broad applicability.

Supporting equipment:

SPE Manifold: used to support SPE column, provide pressure and collect waste liquid, etc. Large capacity injector: Increases the container volume above the column, allowing for a larger sample volume in a single run.

Adapter: Connect the column to other columns or to sample reservoirs, facilitating easier sample loading.



4. Common Terms and Interactions in Solid Phase Extraction (SPE)

Target compound: The compound intended to be separated from a complex sample matrix. Matrix: The environment in which the target compound resides, typically containing many interfering substances. Interfering substances: Compounds that affect the analysis of the target compound or could cause damage to analytical

instruments, usually referring to all compounds in the matrix except for the target compound.

Adsorbent: The packing material in a solid-phase extraction (SPE) column that selectively extracts certain compounds from a sample solution.

Adsorption capacity: The total mass of compounds (including target compounds and some interfering substances) that a given mass of adsorbent can retain under specific conditions.

Selectivity: The ability of the adsorbent to differentiate between the target compound and all other sample components. High selectivity results in better purification by retaining the target compound while excluding other components.

pH: The negative logarithm of the proton (H⁺) concentration in a solution, with smaller values indicating a higher proton concentration.

pKa: The negative logarithm of the acid dissociation constant (Ka). A smaller pKa indicates stronger dissociation of acidic compounds. When the pH of the sample solution equals the pKa, the concentrations of the undissociated and dissociated forms of the compound are equal. For basic compounds, pKa refers to the dissociation constant of the conjugate acid, with larger values indicating a stronger ability to bind protons and strongerbasicity.

Interaction: The forces of attraction or repulsion between two chemical substances (e.g., between the target compound and the adsorbent, or between the target compound and solvent molecules) in a specific chemical environment.

Non-polar interaction: The interaction between non-polar functional groups on the target compound and the non-polar adsorbent. This interaction is most effectively demonstrated in polar solvent environments, particularly in water, hence it is also known as hydrophobic interaction. For example, in a water environment, the interaction between phthalates and C18 is a typical non-polar interaction.

Polar interaction: The interaction between polar functional groups on the target compound and polar functional groups on the adsorbent. This interaction is most effectively observed in weakly polar or non-polarsolvent environments.

Ion interaction: Coulombic forces between ionized functional groups on the target compound and oppositely charged functional groups on the adsorbent.

Secondary interactions: For reversed-phase silica-bonded adsorbents, residual silanol groups on the surface can interact with polar compounds via polar interactions, and dissociated silanol groups can engage in ionic interactions with basic compounds. These interactions are secondary to non-polar interactions and are often undesirable for reversed-phase silica adsorbents. Endcapping techniques are typically used to eliminate these secondary interactions. Activation: Also known as solvation, this process involves introducing a suitable solvent to expose the functional groups on the adsorbent and remove any potential contaminants. For reversed-phase adsorbents, moderately polar solvents (e.g., methanol) are commonly used, while for normal-phase adsorbents, weakly polar or non-polar solvents (e.g., hexane) are preferred. Equilibrium: After activation, a suitable solvent environment is created for sample application by removing the activation solvent. The equilibration solvent is usually the same as the sample solvent. For ion exchange columns, acids are often added to the equilibrating solution if the sample contains basic compounds, while bases are added for acidic compounds. Retention: When the sample solution passes through the adsorbent, compounds are retained if the interaction between the adsorbent and the compounds is stronger than that between the compounds and the solvent. Washing: After sample loading, both interfering substances and the target compound may be retained. A suitable wash solution is introduced to remove as much of the interfering substances as possible without affecting the retention of the target compound. A sample solvent with similar strength to the loading solvent typically does not impact recovery, but a stronger solvent wash can more effectively remove interfering substances. When selecting a washing solution, it is essential to balance recovery rate and purification efficiency.

Penetration: This occurs when the retention capacity of the adsorbent is insufficient or when the mass of the compound exceeds the adsorbent's capacity, resulting in partial or total target compound elution during sample application. This phenomenon is considered an operational error and should be avoided.

5. Interactions in Solid Phase Extraction (SPE)

In the solid phase extraction process, the retention and elution of compounds on the adsorbent are controlled by the interactions between the adsorbent and the compound, as well as between the solvent and the compound. When the interaction between the adsorbent and the compound is stronger than that between the solvent and the compound, the compound is retained by the adsorbent; otherwise, it is eluted. Understanding the interactions between the adsorbent, the compound, and the solvent is crucial for establishing and optimizing solid phase extraction methods. Therefore, this section is a core part of the solid phase extraction technology documentation.

The interactions involved in solid phase extraction can be categorized into four types: Non-polar, polar, ionic exchange, and covalent bonds. Each adsorbent can have more than one type of interaction with the target compound.

Non-Polar Interactions

Non-polar interactions refer to the forces between hydrocarbon groups on the adsorbent and hydrocarbon groups on the target compound. These groups exhibit non-polar or weakly polar characteristics and are only subject to a type of interaction known as "dispersion forces" (a type of Van der Waals force). Since most organic compounds contain varying degrees of non-polar groups, non-polar interactions will retain these compounds on adsorbents with non-polar functional groups. Unmodified silica gel does not exhibit non-polar interactions, but adsorbents obtained by bonding chain functional groups to a silica gel matrix exhibit a degree of non-polarity, making silica gel bonded phases show non-polar adsorbent, and it interacts with target compounds primarily through non-polar interactions. C8 (silica gel bonded with octyl groups) and PH (silica gel bonded with phenyl groups) have weaker non-polar interactions compared to C18, but non-polar interactions are still the primary force for these adsorbents, with other interactions with target compounds being negligible. C2 (silica gel bonded with ethyl groups) and CN (silica gel bonded with cyanoethyl groups) have shorter carbon chains and exhibit both non-polar and polar interactions, but non-polar interactions still predominate. Adsorbents with polar and ionic functional groups (such as NH₂, PSA, SCX, SAX etc.) have strong polar characteristics, making non-polar interactions with target compounds relativelyinsignificant.

For polymer-based adsorbents, such as HLB, MCX, P-SAX, and PWAX, which use polystyrene/divinylbenzene copolymer as the matrix, the phenyl and vinyl groups in the adsorbent are non-polar. Non-polar interactions are a significant force between these adsorbents and target compounds. Adsorbents like MCX, P-SAX, and PWAX also exhibit strong ion-exchange interactions with ionic compounds, which is another important force.

The strength of the interactions between the adsorbent and target compounds is also influenced by the solvent environment. Generally, a strongly polar solvent environment can enhance non-polar interactions between non-polar adsorbents and target compounds. Even if the target compound contains polar groups, its non-polar portions will interact with non-polar adsorbents in a polar environment. Therefore, when using non-polar and weakly polar adsorbents, pure water is the best sample solvent as it increases non-polar interactions, promoting the retention of target compounds. On the other hand, weakly polar organic solvents can dissolve target compounds to some extent, disrupting non-polar interactions between the target compounds and adsorbents. For example, polar solvents like methanol have enough non-polar character to disrupt non-polar interactions between many weakly polar compounds and non-polar adsorbents, leading to the elution of compounds from the adsorbent. For even less polar target compounds, weak or non-polar solvents such as ethyl acetate, tert-butyl methyl ether, or n-hexane are required for elution.

In general, non-polar extraction methods have less selectivity compared to polar or ion-exchange extraction methods, especially when the target compound's structure is similar to the sample matrix components. However, non-polar interactions are very effective for separating compounds with differing structures.

In summary, when retaining target compounds through non-polar interactions (i.e., using reversed-phase solid phase extraction columns), polar solvents (especially pure water) can enhance the retention of these separations and can be chosen as sample and wash solvents. Weakly polar solvents or mixed solvents can disrupt non-polar interactions between target compounds and adsorbents, facilitating the elution of target compounds from non-polaradsorbents.

Polar Interactions

Various adsorbents exhibit polar interactions with the functional groups of target compounds. Polar interactions include hydrogen bonding, dipole-dipole interactions, induced dipole interactions, π - π interactions, and other forms of interaction forces. Polar groups typically contain atoms with significant electronegativity differences, leading to different electron density between these atoms, which imparts polarity to the functional groups. This property enables molecules with polar functional groups to interact with polar functional groups on the adsorbent. Typical polar interacting groups include hydroxyl groups, amino groups, carbonyl groups, thiol groups, double bonds, and groups containing heteroatoms (such as oxygen, nitrogen, fluorine, sulfur, and phosphorus).

Due to the strong polarity of the silica gel matrix (especially the free silanol groups), polar interactions are widespread in silica gel-bonded adsorbents. In non-polar solvents, secondary polar interactions on silica gel-bonded adsorbents are particularly pronounced, with amino and hydroxyl groups being highly sensitive to secondary interactions. Non-polar adsorbents (such as C18, C8, PH, and CH) bonded with non-polar groups are typically used to retain non-polar and weakly polar compounds. The residual silanol groups on the silica gel matrix have undergone end-capping treatment, and such adsorbents are usually operated in polar solvent environments, making secondary interactions in these silica gel-bonded adsorbents very weak. In contrast, in polar silica gel-bonded adsorbents (Silica, NH₂, PSA) and ion-exchange silica gel-bonded adsorbents (SCX, SAX), polar interactions are desired, so no end-capping treatment is needed to suppress secondary interactions.

Hydrogen bonding is one of the most significant forms of polar interaction. The condition for hydrogen bonding is that a hydrogen atom covalently bonded to an electronegative atom X (such as fluorine, chlorine, oxygen, nitrogen) comes close to another electronegative atom Y (such as fluorine, chlorine, oxygen, nitrogen), forming an X-H...Y bond mediated by hydrogen. Hydroxyl or amino groups are primary hydrogen bond donors, and functional groups that can attract hydrogen bond donors (i.e., hydrogen bond acceptors) include those containing oxygen, nitrogen, orsulfur. Non-polar solvents can promote the retention of polar separations on polar adsorbents because non-polar solvent molecules cannot easily disrupt the polar interactions between the adsorbent and the target compound. Conversely, polar solvents can effectively disrupt these polar interactions because polar target compounds are soluble in polar solvents and polar solvents can more effectively compete with the target compounds for adsorption sites on the adsorbent. High ionic concentrations can also disrupt polar interactions. Polar target compounds are often retained on non-polar adsorbents through secondary interactions with the silica gel matrix, but this retention is suppressed by high ionic concentrations. If secondary interactions are required, Tris buffer can be used to enhance this interaction through the adsorbent. In summary, when retaining target compounds through polar interactions, non-polar solvents (especially n-hexane) can enhance the retention of these separations and are suitable for use as sample and wash solvents. Polar solvents and high ionic strength solvents can disrupt the polar interactions between target compounds and adsorbents, facilitating the elution of separations from polar adsorbents. Polar secondary interactions are significant factors in extracting amino or hydroxyl-containing target compounds from non-polar solvents into polaradsorbents.

Ion Interactions

Ion interactions refer to the Coulombic forces occurring between charged target compounds (either positive or negative) and adsorbents with opposite charges. Based on the nature of the ions formed by target compound groups and adsorbent groups, ion exchange interactions can be classified into two types:

A: Cationic Groups (Positive Charge): Organic compounds such as primary, secondary, tertiary amines, and quaternary ammonium compounds, as well as inorganic cations like calcium, sodium, and magnesium, can become cationic. B: Anionic Groups (Negative Charge): Sulfonic acid, carboxylic acid, phosphoric acid, and similar groups can become anionic.

These groups have the potential to form cations or anions, but they are not ions themselves; their ionization depends on the pH of the solvent environment. For effective retention of target compounds via ion exchange, the following conditions must be met: (1) Charge Condition: The pH of the matrix/solvent must result in both the target compound and adsorbent being charged.

(2) Competing lons: The concentration of competingions with the same charge as the target compound should be low in the medium/solution.

To achieve the first condition, it is important to know the pKa values of acidic or basic compounds' conjugate acids. When the environmental pH equals a compound's pKa, 50% of the compound molecules will be charged while the other 50% will be neutral. The relationship with pKa is as follows: When pH is below the pKa, the number of positively charged molecules increases; conversely, it decreases when pH is above the pKa. Similarly, when pH is above the pKa, the number of negatively charged molecules increases, and it decreases when pH is below thepKa. To retain the target compound, the solvent/matrix pH should be in a range where both the target compound and adsorbent are charged. Specifically, the pH should be lower than the pKa of the conjugate acid of basic compounds by at least 2 units and higher than the pKa of acidic compounds by 2 units. At this pH, more than 99% of the target compound will be charged. The pH of the solution during loading should meet this requirement. Conversely, if the pH of the solvent system is above the pKa of acidic compounds or below the pKa of acidic compounds, the target compound's ionic groups will tend towards neutrality, reducing retention. The pH of the elution solvent should meet the requirements to elute ionic target compounds.

Ion strength is also a crucial factor in ion exchange. Ion concentration measures the total concentration of all ions in the solvent/matrix environment. Since ion exchange follows a competitive mechanism, other ions with the same charge in the solvent/matrix will compete with the target compound for ion exchange sites on the adsorbent, affecting the retention of the target compound. Low ion strength enhances retention, while high ion strength weakens it.

Ion exchange adsorbents exhibit strong selectivity for specific ion groups due to the molecular properties of the adsorbent. For example, guaternary ammonium ions (a strong anion-exchange adsorbent) show a 250-fold higher selectivity for citrate compared to acetate. As a result, guaternary ammonium adsorbents equilibrated with citrate have higher retention for anionic target compounds than those equilibrated with acetate. Similarly, for anionic target compounds retained on guaternary ammonium adsorbents, citrate buffer elutes them much more effectively than acetate buffer. Proper use of adsorbent selectivity (for opposite ions) can greatly improve ion exchange extraction.

Due to the presence of unbonded silanol groups on the silica gel matrix, which can partially dissociate protons to form negative charges, all silica gel-bonded phases exhibit ion secondary interactions. These interactions are significant in aqueous environments, with amine groups being the most affected by ion secondary interactions. For example, when amino compounds in water samples are retained on non-polar adsorbents, secondary ion interactions might also play a role. During elution with water/organic solvent mixtures (such as methanol-water), secondary interactions become more pronounced. Despite methanol's ability to disrupt non-polar interactions, active secondary interactions can inhibit the elution of amine compounds. To disrupt secondary adsorption, adjustments in pH (higher pH to neutralize basic compounds; lower pH to neutralize silanol groups) or the addition of competitors (such as diethylamine or triethylamine) in the elution solvent can be used. Competitors will compete with amine compounds for the silanol groups on the adsorbent, breaking secondary adsorption and facilitating the elution of amine compounds. Commonly used silica gel-bonded reversed-phase adsorbents, such as C18, C8, PH, CH, and C2, are end-capped during synthesis, so their ion secondary interactions are minimal. Sometimes, to increase the retention of amine compounds, end-capping is intentionally omitted during the synthesis of silica gel-bonded reversed-phase adsorbents, such as in C18 (nonend-capped).

Polymer-based adsorbents do not contain silanol groups, so they do not exhibit ion secondary interactions. Common ion exchange adsorbents include SCX and SAX. SCX, which is silica gel-bonded with benzenesulfonic acid groups, is very suitable for retaining amine compounds. SAX, which is silica gel-bonded with quaternary ammonium groups, is ideal for retaining compounds with carboxyl and phenolic hydroxyl groups. MCX, PSAX, and PWCX, P-WAX are polymer-based ion exchange reversed-phase adsorbents that combine ion exchange and reversed-phase retention, providing superior purification effects.

To enhance ion exchange and retention of target compounds:

- Maintain the pH of the solvent/matrix between the pKa values of the target compound and adsorbent.
- Use low ion strength in the solvent/matrix.
- Equilibrate the adsorbent with low-selectivity oppositeions.

To promote elution from ion exchange adsorbents:

- Ensure the pH of the solvent/matrix is above the pKa of the conjugate acid of basic compounds or below the pKa of acidic compounds.

- Use high ion strength in thesolvent/matrix.
- Include high-selectivity opposite ions in the solvent/medium.
- Ion secondary interactions are significant for the retention of protonated amine compounds in polar solvents.

Note: Opposite ions are ions with charges opposite to those of the ion groups on the adsorbent.

Operation Method of SPE

Solid Phase Extraction (SPE) is a purification technique that can be categorized into two modes: retaining target compounds and retaining interfering substances.

1. Retaining Target Compounds

The solid-phase extraction mode for retaining target compounds refers to a process where, as the sample solution passes through the adsorbent, the target compound and some interfering substances are retained, while most of the interfering substances are flushed out with the solvent. Then, a wash solution is added to remove the retained interfering substances, and finally, the target compound is eluted with an elution solution. This purification mechanism is the most commonly used method.

In the mode of retaining target compounds, the SPE process involves the following steps: (1) Activation/equilibrium: Activate the adsorbent in the column with an organic solvent to remove any contaminants or interfering substances that may be present in the adsorbent. Then equilibrate the adsorbent with a solution that matches the sample solvent to create a suitable environment for sample loading. (2) Sample loading: Add the sample solution to the column. During this step, the target compound and some interfering substances are retained by the adsorbent, while the rest of the interfering substances are eluted with the sample solvent. (3) Washing: Add a washing solution that has a higher elution strength than the sample solvent but does not elute the target compound. This step cleans the adsorbent by removing the retained interfering substances. (4) Elution: Add an elution solution that can effectively elute the target compound from the adsorbent. Collect the eluate for direct analysis or further processing.

Key Considerations:

Selection of Sample Solvent: The sample solvent must not elute the target compound during the sample loading step. Selection of Washing Solution: The washing solution should maximize the removal of interfering substances without eluting the target compound.

Selection of Elution Solution: The elution solution must be chosen to ensure that it completely elutes the target compound. The success of the SPE method depends significantly on the careful selection of the sample solvent, washing solution, and elution solution. The optimization of these solutions will be discussed in the section on "Optimization of Solid Phase Extraction Methods.



2. Retaining Interfering Substances

The solid-phase extraction mode for retaining interfering substances refers to a process where, as the sample solution passes through the adsorbent, the main interfering substances are retained, while the target compound and some impurities are flushed out with the solvent. By adding an appropriate amount of solvent, the target compound is completely eluted. This mechanism is often used for analyzing multiple pesticide residues in fruits and vegetables, as well as for removing lipophilic interfering substances in ion analysis.

In the mode of retaining interfering substances, the SPE process involves the following steps:

(1) Activation/equilibrium: Use a solution that matches the sample solvent to activate and equilibrate the column, creating a suitable environment for sample loading.

(2) Sample loading: Add the sample solution to the column. During this step, the main interfering substances are retained by the adsorbent, while the target compounds and some impurities flow out with the sample solvent. Collect the eluate.

(3) Washing: Wash the column with a small amount of solvent that matches the sample solvent to remove any residual target compounds from the adsorbent. Collect this eluate and combine it with the eluate collected in the sample loading step. Key Considerations:

Selection of Activation Solution: The activation solution should match the sample solvent to ensure proper equilibration of the adsorbent.

Selection of Sample Solvent: The sample solvent should not interfere with the retention of the main interfering substances. Selection of Washing Solution: The washing solution should also match the sample solvent to effectively remove the target compounds without affecting the retention of the interfering substances.

Selection of Adsorbents and Solvent Systems:

For Polar Interferents: Use polar adsorbents such as silica gel or amino adsorbents. Choose weakly polar to non-polar solvent systems for both the sample solvent and washing solution.

For Weakly Polar or Non-Polar Interferents: Use non-polar adsorbents such as C18 or HLB. Select polar solvent systems, such as pure water, for the sample solvent and washing solution.



General operation method of reversed-phase SPE column



General operation method of normal phase SPE column

Activation/equilibrium: activate or equilibrate with hexane or sample solvent



General operation method of cation exchange SPE column

Activation/equilibrium: When the sample solvent is a sample solvent. When the sample solvent is a polar solvent, first activate with a water-soluble organic

$\mathbf{\Sigma}$

General operation method of anion exchange SPE column

Activation/equilibrium: When the sample solvent is a nonpolar solvent, activate and equilibrate with the

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Washing: Wash with n-hexane, n-hexane containing an

Elution: Elute with a water solution or organic solvent solution with a pH at least two units higher than the Commonly used eluents include ammonia solution and ammonia-methanol solution.

 \Diamond

Washing: Wash sequentially with water and organic

Elution: Elute with a water solution or organic solvent solution with a pH at least two units lower than the pKa of the acidic compound. Commonly used eluents include formic acid solution and formic acid-methanol

 \Diamond

Washing: Wash sequentially with water and organic

SPE INTRODUCTION AND ORDERING INFORMATION

Polymeric SPE: HLB, PS/DVB, MCX, P-SAX, P-WCX, P-WAX, PA

Silica based SPE: C18E, C8, C18, Phenyl, Silica, CN, NH₂, Diol, PSA, SCX, SAX, WCX, PRS

Inorganic SPE: Florisil, Aluminia-N/B/A, Na2SO4, Carb, Acticarbon

Mixed mode: C8/SCX, Carb/NH₂, SAX/PSA, Carb/PSA, C18/CN

Polymeric SPE

SPE

PRODUCTS

Polymer based packing materials have overcome the shortcomings of traditional silica matrix packing materials, and the use proportion of SPE has been increasing gradually.

(1) Broad pH Range: They are suitable for a wide pH range (0-14) and compatible with most organic solvents.

(2) No Active Hydroxyl Groups: The polymer surface lacks active hydroxyl groups, eliminating the secondary adsorption effects that can lead to reduced recovery rates of basic compounds.

(3) High Adsorption Capacity: The polymer matrix has a high adsorption capacity for most organic substances and higher recovery rates, with easy quantitative elution and better reproducibility of analytical results.

(4) Low Detection Limits: The high recovery rate and capacity reduce detection limits and the amount of polymer adsorbent needed.

(5) No Hydrolysis Contamination: The polymer adsorbent does not contaminate expensive extracts due to the hydrolysis of bonded phases.

(6) High Interference Stability: Polymer matrix extraction columns can be re-wet if they accidentally dry out during sample preparation, maintaining their performance without the risk of losing analytes or affecting result reproducibility. Currently available options include HLB, MCX, P-WCX, PWAX, PSAX, and PS/DVB for sample processing.

Different ion groups bonded to the phenyl rings of HLB result in mixed-mode ion exchange reversed-phase adsorbents:

(1) MCX: A mixed-mode strong cation exchange reversed-phase adsorbent, bonded with sulfonic acid groups, combining cation exchange and reversed-phase retention. It is suitable for basic compounds with conjugate acids having pKa values between 2-10, mainly amino compounds.

(2) PSAX: A mixed-mode strong anion exchange reversed-phase adsorbent, bonded with quaternary ammonium groups, combining anion exchange and reversed-phase retention. It is suitable for carboxylic acid compounds with pKa values between 2-8.

(3) P-WCX: A mixed-mode weak cation exchange reversed-phase adsorbent, bonded with carboxyl groups, combining weak cation exchange and reversed-phase retention. It is suitable for strongly basic compounds with conjugate acids having pKa values greater than 10, such as compounds containing guaternary ammonium groups. (4) PWAX: A mixed-mode weak anion exchange reversed-phase adsorbent, bonded with piperazine groups, combining weak anion exchange and reversed-phase retention. It is suitable for strongly acidic compounds with pKa values less than 1, such as compounds containing sulfonic acid or phosphoric acidgroups.

1. HLB

HLB is a monodisperse hydrophilic-hydrophobic balance reversed-phase adsorbent, which is a polymer adsorbent modified through surface modification to introduce polar functional groups. It is used for the separation of both polar and non-polar substances, with an adsorption capacity 3-10 times greater than that of C18-bonded silica stationary phases. It is effective for the extraction and separation of drugs such as atropine, ibuprofen, fenoprofen, indomethacin, caffeine, the obro-mine, and diazepam. It is comparable to Waters' Oasis® HLB solid-phase extraction columns.

Technical parameters

| Matrix | Divinyl benzene polymer |
|---------------------|-----------------------------|
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Phenyl, vinyl, pyrrolidone |
| Retention mechanism | Reversed-phase retention |



1. Food Safety Testing: Analysis of drug residues in animal samples, such as tetracyclines, chloramphenicol, sulfonamides, avermectin, quinolones, macrolide antibiotics, nitrofurans, and their metabolites; pesticide residue analysis in plant samples; detection of food additives in food, such as dimethyl fumarate, neotame, and sucralose.

2. Environmental Monitoring: Analysis of PAHs, PAEs, phenolic compounds, bisphenol A, and triazine herbicides in water and soil.

3. Biological Samples: Analysis of drugs in blood and urine, such as tetracyclines, cocaine and its metabolites, morphine and its metabolites, barbiturates, tricyclic antidepressants, andranitidine.

Ordering information of HLB

| Part No. | Description | Part No. | Description |
|--------------|---------------------------|--------------|--------------------------|
| YZSC1001-HLB | HLB, 100mg/1ml, 100pcs/pk | YZSC5003-HLB | HLB, 500mg/3ml, 50pcs/pk |
| YZSC603-HLB | HLB, 60mg/3ml, 50pcs/pk | YZSC1506-HLB | HLB, 150mg/6ml, 30pcs/pk |
| YZSC1003-HLB | HLB, 100mg/3ml, 50pcs/pk | YZSC2006-HLB | HLB, 200mg/6ml, 30pcs/pk |
| YZSC2003-HLB | HLB, 200mg/3ml, 50pcs/pk | YZSC5006-HLB | HLB, 500mg/6ml, 30pcs/pk |

MAX HLB HLB Carb Carb MCX MCX Florisit

2. PS/DVB

PS/DVB is a polymer adsorbent based on highly cross-linked polystyrene/divinylbenzene copolymer. It features an extremely high specific surface area (800 m²/g) and very high adsorption capacity. It is used for the rapid adsorption and separa- tion of hydrophobic substances such as phenols, surfactants, bromofluorocarbons, antibiotics, amino acids, and peptides. It can extract polar compounds that are not sufficiently retained by C18 and C8 stationary phases. It is equivalent to Bond Elute LMS and Bond Elute PPL.

Technical parameters

| Matrix | Divinyl benzene polymer |
|---------------------|-----------------------------|
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Phenyl, vinyl |
| Retention mechanism | Reversed-phase retention |
| | |

Application

Food Safety Testing: Analysis of drug residues in animal samples; analysis of pesticide residues in plant samples; analysis of food additives in seasonings and processed foods; analysis of antioxidants in vegetableoils.
Environmental Monitoring: Analysis of phenolic compounds.
Biological Samples: Analysis of drugs in blood andurine.

Ordering information of PS/DVB

| P/N | Description | P/N | Description |
|-------------|------------------------------|--------------|-----------------------------|
| YZSC301-PD | PS/DVB, 30mg/1ml, 100pcs/pk | YZSC5003-PD | PS/DVB, 500mg/3ml, 50pcs/pk |
| YZSC1001-PD | PS/DVB, 100mg/1ml, 100pcs/pk | YZSC1006-PD | PS/DVB, 100mg/6ml, 30pcs/pk |
| YZSC603-PD | PS/DVB, 60mg/3ml, 50pcs/pk | YZSC1506-PD | PS/DVB, 150mg/6ml, 30pcs/pk |
| YZSC1003-PD | PS/DVB, 100mg/3ml, 50pcs/pk | YZSC2006-PD | PS/DVB, 200mg/6ml, 30pcs/pk |
| YZSC1503-PD | PS/DVB, 150mg/3ml, 50pcs/pk | YZSC5006-PD | PS/DVB, 500mg/6ml, 30pcs/pk |
| YZSC2003-PD | PS/DVB, 200mg/3ml, 50pcs/pk | YZSC10006-PD | PS/DVB, 1g/6ml, 30pcs/pk |
| YZSC2503-PD | PS/DVB, 250mg/3ml, 50pcs/pk | | |

e size: 80Å Surface area: 800-1000m²

3. MCX

Polymer matrix adsorbents often exhibit mixed retention mechanisms, and this is also true for Yicozoo polymer matrix SPE products. Yicozoo MCX is a mixed adsorbent with strong cation exchange functionality and hydrophobic interaction as a reversed-phase chromatography stationary phase. It is bonded with strong acidic sulfonic acid functional groups, selectively retaining both basic and neutral compounds. It is commonly used for the extraction of basic substances such as melamine, amphetamine, diphenhydramine, and phenylcyclohexylamine. It is equivalent to Waters' Oasis® MCX.

Technical parameters

| Matrix | Divinyl benzene polymer |
|---------------------|--|
| Parameter | Particle size: 40-60µm Pore size: 80Å Surface area: 800-1000m ² |
| Function groups | Pyrrolidone, benzene sulfonic acid functional groups, phenyl, vinyl |
| Retention mechanism | Reversed-phase retention and strong cationic mechanism |

General operation method of MCX (Take MCX, 60mg/3mL as an example)



Application

1. Food Safety Testing: Analysis of melamine; analysis of basic drug residues in animal samples, such as metronidazole, nitroimidazoles, sulfonamides, and clenbuterol; analysis of basic pesticides in vegetables, fruits, and fruit juices, such as carbendazim and thiophanate-methyl.

2. Biological Samples: Analysis of basic drugs in blood andurine.

Ordering information of MCX

| P/N | Description | P/N | Description |
|--------------|---------------------------|--------------|--------------------------|
| YZSC301-MCX | MCX, 30mg/1ml, 100pcs/pk | YZSC2003-MCX | MCX, 200mg/3ml, 50pcs/pk |
| YZSC1001-MCX | MCX, 100mg/1ml, 100pcs/pk | YZSC5003-MCX | MCX, 500mg/3ml, 50pcs/pk |
| YZSC603-MCX | MCX, 60mg/3ml, 50pcs/pk | YZSC1506-MCX | MCX, 150mg/6ml, 30pcs/pk |
| YZSC1003-MCX | MCX, 100mg/3ml, 50pcs/pk | YZSC2006-MCX | MCX, 200mg/6ml, 30pcs/pk |
| YZSC1503-MCX | MCX, 150mg/3ml, 50pcs/pk | YZSC5006-MCX | MCX, 500mg/6ml, 30pcs/pk |

4. P-SAX

P-SAX is a mixed-phase adsorbent with strong anion exchange functionality and reverse-phase hydrophobic proper- ties. It contains guaternary ammonium functional groups and is typically used for the separation and purification of acidic substances from basic and neutral impurities, such as phosphoric acid, estrogen, adenine, and nucleosides. The polymer-based P-SAX is resistant to many organic solvents and is stable in aqueous solutions across a pH range of 0-14.

Technical parameters

| Matrix | Divinyl benzene polymer |
|---------------------|-----------------------------|
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Pyrrolidone, quaternary an |
| Retention mechanism | Reversed-phase retention |

Application

1. Detection and analysis of food additives in foods and seasonings, such as benzoic acid, sorbic acid, dehydroacetic acid, vanillin, methylvanillin, and ethylvanillin.

2. Detection and analysis of mycotoxins in food, such as patulin, citrinin, ochratoxin A, and mycophenolic acid. 3. Analysis of components in herbs, such as the detection of 6-gingerol, curcumin, and piperine inturmeric.

4. Detection of caffeine in spirits.

5. Analysis of pesticide residues in food, such as sodium pentachlorophenate and cyhalothrin.

Ordering information of P-SAX

| P/N | Description | P/N | Description |
|---------------|----------------------------|---------------|----------------------------|
| YZSC603-PSAX | P-SAX, 60mg/3ml, 50pcs/pk | YZSC1506-PSAX | P-SAX, 150mg/6ml, 30pcs/pk |
| YZSC2003-PSAX | P-SAX, 200mg/3ml, 50pcs/pk | YZSC2006-PSAX | P-SAX, 200mg/6ml, 30pcs/pk |
| YZSC606-PSAX | P-SAX, 60mg/6ml, 30pcs/pk | YZSC5006-PSAX | P-SAX, 500mg/6ml, 30pcs/pk |

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| e size: 80Å Surface area: 800-1000m ² |
|--|
| nmonium functional groups, phenyl, vinyl |
| and strong anionic mechanism |

5. P-WCX

P-WCX column are packed with a polymer-based weak cation exchange adsorbent. The retention mechanism combines weak cation exchange with reversed-phase retention. The P-WCX column packing is bonded with carboxylic acid groups, making it suitable for the separation and purification of strongly basic compounds. The divinylbenzene polymer matrix ensures that the column maintains stable performance even under high pH conditions.

Technical parameters

| Matrix | Divinyl benzene polymer |
|---------------------|--|
| Parameter | Particle size: 40-60µm Pore size: 80Å Surface area: 800-1000m ² |
| Function groups | Pyrrolidone, carboxylic, phenyl, vinyl |
| Retention mechanism | Reversed-phase retention and weak cationic mechanism |

General operation method of P-WCX (Take P-WCX, 60mg/3ml as an example)



Application

Separate and purify strongly basic compounds, such as those with quaternary ammonium groups.

Ordering information of P-WCX

| P/N | Description | P/N | Description |
|--------------|---------------------------|---------------|----------------------------|
| YZSC301-PWCX | P-WCX 30mg/1ml, 100pcs/pk | YZSC5003-PWCX | P-WCX, 500mg/3ml, 50pcs/pk |
| YZSC603-PWCX | P-WCX 60mg/3ml, 50pcs/pk | YZSC5006-PWCX | P-WCX, 500mg/6ml, 30pcs/pk |

6. P-WAX

P-WAX column is a polymer-based weak anion exchange column. The packing material exhibits both weak anion exchange and reversed-phase retention mechanisms. The piperazine group is bonded to the phenyl ring of the HLB packing, making it suitable for the separation and purification of strongly acidic substances. The divinylbenzene polymer ensures stable performance across the pH range of 0-14.

Technical parameters

| Matrix | Divinyl benzene polymer |
|---------------------|------------------------------|
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Pyrrolidinone, piperazine, p |
| Retention mechanism | Reversed-phase retention |

Application

1. Separation and purification of strongly acidic compounds, such as compounds containing sulfonic acidgroups. 2. Detection and analysis of artificial pigment in food, better recovery effect of erythrosine than that of PA column.

Ordering information of P-WAX

| P/N | Description | P/N | Description |
|---------------|----------------------------|---------------|----------------------------|
| YZSC301-PWAX | P-WAX, 30mg/1ml, 100pcs/pk | YZSC5003-PWAX | P-WAX, 500mg/3ml, 50pcs/pk |
| YZSC603-PWAX | P-WAX, 60mg/3ml, 50pcs/pk | YZSC1506-PWAX | P-WAX, 150mg/6ml, 30pcs/pk |
| YZSC1503-PWAX | P-WAX, 150mg/3ml, 50pcs/pk | YZSC2006-PWAX | P-WAX, 200mg/6ml, 30pcs/pk |
| YZSC2003-PWAX | P-WAX, 200mg/3ml, 50pcs/pk | YZSC5006-PWAX | P-WAX, 500mg/6ml, 30pcs/pk |

7. PA

Adsorbent of PA column is polyamide, a polymeric compound containing amide groups in the molecular backbone repeating unit, commonly known as "nylon". Amide groups can form hydrogen bonds with polar compounds and have a good adsorption and retention effect on polar compounds. Both column and packing materials are suitable for the determination of synthetic colorants in food such as lemon yellow, new red, amaranth red, carmine red, sunset yellow, bright blue and so on or the removal of pigment in samples.

Application

Determination of synthetic colorants in food. Leather and fur - Determination of hexavalent chromium content - Spectrophotometric method.

Ordering information of PA

| P/N | Description | P/N | Description |
|-------------|------------------------|---------------|------------------------|
| YZSC1001-PA | PA,100mg/1mL,100pcs/pk | YZSC5006-PA | PA,500mg/6mL,30pcs/pk |
| YZSC603-PA | PA,60mg/3mL,50pcs/pk | YZSC10006-PA | PA,1g/6mL,30pcs/pk |
| YZSC1003-PA | PA,100mg/3mL,50pcs/pk | YZSC200012-PA | PA,2g/12mL,20pcs/pk |
| YZSC1503-PA | PA,150mg/3mL,50pcs/pk | YZSC300012-PA | PA,3g/12mL,20pcs/pk |
| YZSC2003-PA | PA,200mg/3mL,50pcs/pk | YZSC400060-PA | PA,4g/60mL,10pcs/pk |
| YZSC5003-PA | PA,500mg/3mL,50pcs/pk | YZSS010-PA | PA Sorbent,10g/bottle |
| YZSC2006-PA | PA,200mg/6mL,30pcs/pk | YZSS100-PA | PA Sorbent,100g/bottle |

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e size: 80Å Surface area: 800-1000m²

phenyl, vinyl

and weak anionic mechanism

Silica Based SPE

Silica-based SPE products use high-quality, high-purity amorphous silica with an average particle size of 45 µm, an average pore size of 60 Å, a pore volume of 0.80 cm³/g, and a specific surface area of 480 m²/g. This type of silica provides moderate resistance and flow rate during extraction, making it an ideal choice. Additionally, the bonding silica uses Yicozoo's unique surface treatment process, and the bonding phase employs a more stable trivalent bonding method, ensuring consistent extraction and recovery rates for analytes. Silica or bonded silica remains the most commonly used adsorbent in SPE, with a pH range of 2-7.5. Silica-based adsorbents offer a wide variety of options. Yicozoo SPE includes 13 types of silica-based phases: C18E (end-capped), C18 (non-end-capped), C8, Phenyl, CN, NH₂, PSA, SCX, SAX, WAX, PRS, Silica, and Diol.

General properties of silica-based adsorbent:

1. The functional groups bonded to the silica surface primarily determine the retention of target compounds. Depending on the type of bonded functional groups, the retention mechanisms include reversed-phase retention, normal-phase retention, and ion-exchange retention.

2. Silica-bonded adsorbents are stable within a pH range of 2-7.5.

3. Silica-bonded adsorbents exhibit rigidity and do not shrink or expand during solvent transformation, achieving equilibrium quickly in new solvents.

4. The silica matrix consists of spherical silica with a particle size of 40-63 µm, with uniform particle size and smooth surface, allowing solvents to pass through even without pressure. The characteristic pore size of silica-bonded adsorbents is approximately 60 Å, suitable for compounds with a molecular weight less than 15,000. Reversed-phase adsorbents are end-capped, while normal-phase and ion-exchange adsorbents are not.

Technical parameters

| Structural Formula | |
|---------------------|-----------------------------|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore |
| Function groups | C18 alkyl chain |
| Endcapped | Yes |
| Carbon loading | 17% |
| Retention mechanism | Reversed-phase retention |

Application

1. Extraction of Organic Contaminants in Water: PAHs, PAEs, PCBs, phenolic substances, microcystin toxins, and pesticide residues (such as abamectin, naphthalene, atrazine, etc.). 2. Life Sciences: Extraction of drugs and their metabolites from plasma, serum, and urine; extraction of pesticide and veterinary drug residues from food. 3. Extraction of Plant and Animal Components: Essential oils, fat-soluble vitamins, water-soluble vitamins, carbohydrates,

organic acids, steroids, and desalting of biomolecules.

4. Food Safety: Detection and analysis of food additives such as neotame; detection and analysis of contaminants in food contact materials and products, such as acrylamide and bisphenol A; pesticide residue analysis in food, such as glyphosate and abamectin.

Ordering information of C18E

| P/N | Description | P/N | Description |
|---------------|----------------------------|------------------|---------------------------|
| YZSC301-C18E | C18E, 30mg/1ml, 100pcs/pk | YZSC5003-C18E | C18E, 500mg/3ml, 50pcspk |
| YZSC1001-C18E | C18E, 100mg/1ml, 100pcs/pk | YZSC3006-C18E | C18E, 300mg/6ml, 30pcs/pk |
| YZSC303-C18E | C18E, 30mg/3ml, 50pcs/pk | YZSC5006-C18E | C18E, 500mg/6ml, 30pcs/pk |
| YZSC603-C18E | C18E, 60mg/3ml, 50pcs/pk | YZSC10006-C18E | C18E, 1g/6ml, 30pcs/pk |
| YZSC1003-C18E | C18E, 100mg/3ml, 50pcs/pk | YZSC200012-C18E | C18E, 2g/12ml, 20pcs/pk |
| YZSC1503-C18E | C18E, 150mg/3ml, 50pcs/pk | YZSC500030-C18E | C18E, 5g/30ml, 10pcs/pk |
| YZSC2003-C18E | C18E, 200mg/3ml, 50pcs/pk | YZSC500060-C18E | C18E, 5g/60ml, 10pcs/pk |
| YZSC3003-C18E | C18E, 300mg/3ml, 50pcs/pk | YZSC1000060-C18E | C18E, 10g/60ml, 10pcs/pk |

1. C18E

C18E is an end-capped C18 adsorbent, known for being the most hydrophobic silica-based adsorbent. It exhibits excellent strong retention characteristics for non-polar compounds and retains most organic substances, making it the most widely used SPE adsorbent. Since C18E retains most organic substances in aqueous matrices, it has minimal selectivity and is commonly used for processing samples with various structures or significant structural differences. Additionally, because C18E does not retain salts at all, it can often replace ion-exchange columns for desalting small molecules and some medium-sized molecules.

volume: 0.80 cm³/g Pore size: 60Å Surface area: 480m²/g

2. C8

C8 adsorbent is similar to C18 in terms of adsorption properties, relying mainly on non-polar interactions. However, because the C8 alkyl chain is shorter than the C18 alkyl chain, its retention of non-polar compounds is weaker compared to C18, which helps in eluting samples with excessively strong non-polar adsorption. C8 columns can simultaneously extract both lipidsoluble and water-soluble vitamins from plasma and are also commonly used for desalting biomolecular samples. Yicozoo C8 is a commonly used non-polar adsorbent. For basic analytes, using C8 adsorbent can increase extraction efficiency and improve recovery rates.

Technical parameters

| Structural Formula | |
|---------------------|---|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore volume: 0.80 cm ³ /g Pore size: 60Å Surface area: 480m ² /g |
| Function groups | Octane group |
| Endcapped | Yes |
| Carbon loading | 12% |
| Retention mechanism | Reversed-phase retention |

Application

1. Extraction of Organic Contaminants in Water: PAHs, PAEs, PCBs, pesticides, herbicides, phenolic compounds, etc.

2. Life Sciences: Extraction of drugs and their metabolites from plasma, serum, and urine.

3. Pesticide and Veterinary Drug Residue Extraction inFoods.

4. Extraction of Plant and Animal Components: Aromatic oils, fat-soluble vitamins, water-soluble vitamins, carbohydrates, organic acids, steroids, etc.

5. Desalting of Biomolecules.

Ordering information of C8

| P/N | Description | P/N | Description |
|-------------|--------------------------|---------------|-------------------------|
| YZSC1001-C8 | C8, 100mg/1ml, 100pcs/pk | YZSC5003-C8 | C8, 500mg/3ml, 50pcs/pk |
| YZSC1003-C8 | C8, 100mg/3ml, 50pcs/pk | YZSC5006-C8 | C8, 500mg/6ml, 30pcs/pk |
| YZSC1503-C8 | C8, 150mg/3ml, 50pcs/pk | YZSC10006-C8 | C8, 1g/6ml, 30pcs/pk |
| YZSC2003-C8 | C8, 200mg/3ml, 50pcs/pk | YZSC200012-C8 | C8, 2g/12ml, 20pcs/pk |

3. C18

C18 is an unendcapped C18 adsorbent. The residual silicon hydroxyl groups on its surface provide additional polar interactions, allowing the hydrophobic adsorbent to make closer contact with more polar extracts, enhancing its retention capability for basic and polar substances. Compared to the endcapped C18E, Yicozoo C18 is a general-purpose adsorbent for the extraction of both polar and non-polar compounds. Yicozoo C18 has a carbon content of approximately 17%, an average particle size of 45 μ m, an average pore size of 60 Å, a pore volume of 0.80 cm³/g, and a specific surface area of 480 m²/g.

Technical parameters

| Structural Formula | Он |
|---------------------|------------------------------|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore |
| Function groups | C18 alkyl chain, silicon hyd |
| Endcapped | No |
| Carbon loading | 17% |
| Retention mechanism | Reversed-phase retention |
| | |

Application

1. Similar to C18E, but with enhanced retention for polar compounds.

2. Detection of agricultural residues in water, such as atrazine, simazine, etc.

3. Detection of food additives in baked food and processed food, such as acesulfame potassium, saccharin sodium, aspartame, etc. 4. Detection of food additives in beverages, such as acesulfame potassium, caffeine,etc.

Ordering information of C18

| P/N | Description | P/N | Description |
|--------------|---------------------------|----------------|--------------------------|
| YZSC1001-C18 | C18, 100mg/1ml, 100pcs/pk | YZSC5003-C18 | C18, 500mg/3ml, 50pcs/pk |
| YZSC603-C18 | C18, 60mg/3ml, 50pcs/pk | YZSC5006-C18 | C18, 500mg/6ml, 30pcs/pk |
| YZSC1003-C18 | C18, 100mg/3ml, 50pcs/pk | YZSC10006-C18 | C18, 1g/6ml, 30pcs/pk |
| YZSC1503-C18 | C18, 150mg/3ml, 50pcs/pk | YZSC100012-C18 | C18, 1g/12ml, 20pcs/pk |
| YZSC2003-C18 | C18, 200mg/3ml, 50pcs/pk | | |

4. Phenyl

Phenyl is a phenyl-bonded SPE stationary phase. The SPE column packing of this material enhances retention of basic compounds through the unique π - π polar interactions of the phenyl ring. When extracting both aromatic and non-aromatic compounds, Phenyl exhibits different selectivity compared to reverse-phase stationary phases like C18 and C8. The silica gel used in Phenyl has an average particle size of 45 µm, a pore size of 60 Å, a pore volume of 0.80 cm³/g, and a specific surface area of 480 m²/g.

e volume: 0.80 cm³/g Pore size: 60Å Surface area: 480m²/g

droxyl

Technical parameters

| Structural Formula | |
|---------------------|---|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore volume: 0.80 cm ³ /g Pore size: 60Å Surface area: 480m ² /g |
| Function groups | Phenyl group |
| Endcapped | Yes |
| Carbon loading | 10% |
| Retention mechanism | Reversed-phase retention |

Application

1. Extraction of Organic Contaminants in Water: PAHs, PAEs, PCBs, pesticides, herbicides, phenolic substances, etc.

- 2. Life Sciences: Extraction of drugs and their metabolites from plasma, serum, and urine.
- 3. Pesticide and Veterinary Drug Residue in Food: Extraction of pesticide and veterinary drug residues.

4. Extraction of Plant and Animal Components: Essential oils, fat-soluble vitamins, water-soluble vitamins, carbohydrates, organic acids, steroids, etc.

Ordering information of Phenyl

| P/N | Description | P/N | Description |
|-----------------|------------------------------|-------------------|-----------------------------|
| YZSC1001-Phenyl | Phenyl, 100mg/1ml, 100pcs/pk | YZSC5003- Phenyl | Phenyl, 500mg/3ml, 50pcs/pk |
| YZSC1003-Phenyl | Phenyl, 100mg/3ml, 50pcs/pk | YZSC5006- Phenyl | Phenyl, 500mg/6ml, 30pcs/pk |
| YZSC1503-Phenyl | Phenyl, 150mg/3ml, 50pcs/pk | YZSC10006-Phenyl | Phenyl, 1g/6ml, 30pcs/pk |
| YZSC2003-Phenyl | Phenyl, 200mg/3ml, 50pcs/pk | YZSC200012-Phenyl | Phenyl, 2g/12ml, 20pcs/pk |

5. Silica

Silica is an unbonded active silica normal-phase adsorbent, which is weakly acidic and has strong polarity. The reten-tion of target compounds is mainly achieved through hydrogen bonding. The silica gel's surface silanol groups can ionize, and under moderate pH conditions, it functions similarly to a weak cation exchange adsorbent.

Technical parameters

| Structural Formula | О—он |
|---------------------|---|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore volume: 0.80 cm ³ /g Pore size: 60Å Surface area: 480m ² /g |
| Function groups | Silicon hydroxyl |
| Endcapped | No |
| Retention mechanism | Normal phase retention |

Application

1. Extraction of compounds with polar groups in lipid samples. 2. Adsorption of interfering substance in extract during pesticide residue analysis.

Ordering information of Silica

| P/N | Description | P/N | Description |
|-----------------|------------------------------|-------------------|-----------------------------|
| YZSC1001-Silica | Silica, 100mg/1ml, 100pcs/pk | YZSC5006-Silica | Silica, 500mg/6ml, 30pcs/pk |
| YZSC2003-Silica | Silica, 200mg/3ml, 50pcs/pk | YZSC10006-Silica | Silica, 1g/6ml, 30pcs/pk |
| YZSC5003-Silica | Silica, 500mg/3ml, 50pcs/pk | YZSC200010-Silica | Silica, 2g/10ml, 20pcs/pk |
| YZSS100-Silica | Silica Sorbent, 100g/bottle | YZSS010-Silica | Silica Sorbent, 10g/bottle |

6. CN

CN is a cyano-polar bonded adsorbent that combines both polar and non-polar interactions. It can be used as a non-polar adsorbent to simultaneously extract both polar and non-polar substances from aqueous samples. It can also extract polar substances from solvents that are relatively less polar.

Technical parameters

| Structural Formula | <u></u> см |
|---------------------|-----------------------------|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Cyano |
| Endcapped | Yes |
| Carbon loading | 6.5% |
| Retention mechanism | Reversed-phase or normal |

Application

Detection of pesticides, drugs and their metabolites in water samples.

Ordering information of CN

| P/N | Description | P/N | Description |
|-------------|--------------------------|---------------|-------------------------|
| YZSC1001-CN | CN, 100mg/1ml, 100pcs/pk | YZSC5003-CN | CN, 500mg/3ml, 50pcs/pk |
| YZSC1003-CN | CN, 100mg/3ml, 50pcs/pk | YZSC5006-CN | CN, 500mg/6ml, 30pcs/pk |
| YZSC1503-CN | CN, 150mg/3ml, 50pcs/pk | YZSC10006-CN | CN, 1g/6ml, 30pcs/pk |
| YZSC2003-CN | CN, 200mg/3ml, 50pcs/pk | YZSC200012-CN | CN, 2g/12ml, 20pcs/pk |

volume: 0.80 cm³/g Pore size: 60Å Surface area: 480m²/g

phase retention

7. NH₂

 NH_2 is an aminopropyl bonded silica adsorbent that can function both as a polar adsorbent and as a weak anion exchange adsorbent. When activated with non-polar solvents such as n-hexane, it can form hydrogen bonds with molecules containing - OH, -NH, or -SH groups. In aqueous environments with a pH < 7.8, it can act as a weak anion exchange adsorbent.

Technical parameters

| Structural Formula | NH2 |
|---------------------|---|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60 μ m Pore volume: 0.80 cm ³ /g Pore size: 60Å Surface area: 480m ² /g |
| Function groups | Aminopropyl |
| Endcapped | No |
| Carbon loading | 3.5% |
| Retention mechanism | Normal phase retention or weak anion exchange |

Application

1. Can be used to separate structuralisomers.

2. Extraction of compounds with polar groups in lipid samples, such as thiomersal in skin care products.

3. Used in agricultural residue analysis to remove polar compounds (such as carbohydrates and pigments), organic acids, phenols, etc.

4. Detection of synthetic colorants in beverages, condiments and processed meats, such as acid orange II, etc.

Ordering information of NH₂

| P/N | Description | P/N | Description |
|----------------|--|----------------|---------------------------------------|
| YZSC1001-NH2 | NH ₂ , 100mg/1ml, 100pcs/pk | YZSC5003-NH2 | NH ₂ , 500mg/3mL, 50pcs/pk |
| YZSC1003-NH2 | NH ₂ , 100mg/3ml, 50pcs/pk | YZSC5006-NH2 | NH ₂ , 500mg/6mL, 30pcs/pk |
| YZSC1503-NH2 | NH ₂ , 150mg/3ml, 50pcs/pk | YZSC10006-NH2 | NH ₂ , 1g/6mL, 30pcs/pk |
| YZSC2003-NH2 | NH ₂ , 200mg/3ml, 50pcs/pk | YZSC200012-NH2 | NH ₂ , 2g/12mL, 20pcs/pk |
| YZSC500030-NH2 | NH ₂ , 5g/30ml, 10pcs/pk | | |

8. Diol

Diol is a silica-based SPE polar adsorbent with bonded diol groups. Depending on the activation conditions and sample matrix, it can also exhibit weak non-polar interactions, allowing it to extract non-polar substances from aqueous samples. In most cases, it is used as a polar adsorbent, similar to unbonded silica in its polar interactions, extracting polar molecules from non-polar solvents. It is useful for separating isomers and other structurally similar compounds.

Technical parameters

| Structural Formula | ОН |
|---------------------|-----------------------------|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Diol |
| Endcapped | No |
| Retention mechanism | Normal phase retention |

Application

Separation and purification of polar target compounds or compounds with moderate polarity.

Ordering information of Diol

| P/N | Description | P/N | Description |
|---------------|----------------------------|----------------|---------------------------|
| YZSC1001-Diol | Diol, 100mg/1ml, 100pcs/pk | YZSC5003-Diol | Diol, 500mg/3ml, 50pcs/pk |
| YZSC1003-Diol | Diol, 100mg/3ml, 50pcs/pk | YZSC5006-Diol | Diol, 500mg/6ml, 30pcs/pk |
| YZSC1503-Diol | Diol, 150mg/3ml, 50pcs/pk | YZSC10006-Diol | Diol, 1g/6ml, 30pcs/pk |
| YZSC2003-Diol | Diol, 200mg/3ml, 50pcs/pk | | |

9. **PSA**

PSA is an adsorbent similar to NH₂. PSA contains two amino groups with pKa values of 10.1 and 10.9. It has a stronger ion exchange capability compared to NH₂ solid phase extraction columns. PSA is generally used in anionic exchange retention mode. Its packing material matrix is silica with an average particle size of 45 μ m, pore size of 60 Å, pore volume of 0.80 cm³/g, and a specific surface area of 480 m²/g.

Technical parameters

| Structural Formula | NH |
|---------------------|-----------------------------|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Ethylenediamine |
| Endcapped | No |
| Carbon loading | 7% |
| Retention mechanism | Normal phase retention or |

volume: 0.80 cm³/g Pore size: 60Å Surface area: 480m²/g

volume: 0.80 cm³/g Pore size: 60Å Surface area: 480m²/g

weak anion exchange

1. Separation of Structural Isomers: Can be used to separate compounds with similar structures, such as isomers.

2. Extraction of Polar Compounds in Lipid Samples: Suitable for extracting compounds with polar functional groups from lipid samples.

Removal of Polar Compounds in Pesticide Residue Analysis: Used to remove polar compounds (such as carbohydrates, pigments), organic acids, phenols, etc., from extraction solutions. For example, detecting acid orange II in processed meat products.
Pesticide Residue Detection in Vegetables: Applied for analyzing pesticide residues in vegetables, such as imidacloprid, difenoconazole, methomyl, cyromazine, and chlorpyrifos.

Ordering information of PSA

| P/N | Description | P/N | Description |
|--------------|---------------------------|----------------|--------------------------|
| YZSC1001-PSA | PSA, 100mg/1ml, 100pcs/pk | YZSC1506-PSA | PSA, 150mg/6ml, 30pcs/pk |
| YZSC1003-PSA | PSA, 100mg/3ml, 50pcs/pk | YZSC2006-PSA | PSA, 200mg/6ml, 30pcs/pk |
| YZSC1503-PSA | PSA, 150mg/3ml, 50pcs/pk | YZSC5006-PSA | PSA, 500mg/6ml, 30pcs/pk |
| YZSC2003-PSA | PSA, 200mg/3ml, 50pcs/pk | YZSC10006-PSA | PSA, 1g/6ml, 30pcs/pk |
| YZSC5003-PSA | PSA, 500mg/3ml, 50pcs/pk | YZSC200012-PSA | PSA, 2g/12ml, 20pcs/pk |

10. SCX

SCX (Strong Cation Exchange) is a strong cation exchange extraction column with a silica gel matrix, bonded with benzenesulfonic acid groups.

Technical parameters

| Structural Formula | С С С С С С С С С С С С С С С С С С С |
|---------------------|---|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore volume: 0.80 cm ³ /g Pore size: 60Å Surface area: 480m ² /g |
| Function groups | Benzene sulfonic acid group |
| Endcapped | No |
| Carbon loading | 2% |
| Retention mechanism | Normal phase retention or cation exchange |

Application

Purification of alkaline compounds in aqueous samples, biological fluids and organic phases and detection of niacin and niacinamide in skin care products and agricultural residues in vegetables, such as cyromazine.

Ordering information of SCX

| P/N | Description | P/N | Description |
|--------------|---------------------------|---------------|--------------------------|
| YZSC1001-SCX | SCX, 100mg/1ml, 100pcs/pk | YZSC2003-SCX | SCX, 200mg/3ml, 50pcs/pk |
| YZSC603-SCX | SCX, 60mg/3ml, 50pcs/pk | YZSC5003-SCX | SCX, 500mg/3ml, 50pcs/pk |
| YZSC1003-SCX | SCX, 100mg/3ml, 50pcs/pk | YZSC5006-SCX | SCX, 500mg/6ml, 30pcs/pk |
| YZSC1503-SCX | SCX, 150mg/3ml, 50pcs/pk | YZSC10006-SCX | SCX, 1g/6ml, 30pcs/pk |

11. SAX

SAX (Strong Anion Exchange) is a strong anion exchange extraction column with quaternary ammonium functional groups bonded to the silica surface. It is primarily used for the adsorption and enrichment of weakly acidic target substances, such as organic acids. It is mainly used to concentrate negatively charged target substances from aqueous or non-aqueous solutions.

Technical parameters

| Structural Formula | |
|---------------------|-----------------------------|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Quaternary ammonium gro |
| Endcapped | No |
| Carbon loading | 7.5% |
| Retention mechanism | Normal phase retention or |

Application

Purification of alkaline compounds from aqueous samples, biological fluids and organic phases.

Ordering information of SAX

| P/N | Description | P/N | Description |
|---------------|---------------------------|--------------|--------------------------|
| YZSC1001-SAX | SAX, 100mg/1ml, 100pcs/pk | YZSC2003-SAX | SAX, 200mg/3ml, 50pcs/pk |
| YZSC603-SAX | SAX, 60mg/3ml, 50pcs/pk | YZSC5003-SAX | SAX, 500mg/3ml, 50pcs/pk |
| YZSC1003-SAX | SAX, 100mg/3ml, 50pcs/pk | YZSC606-SAX | SAX, 60mg/6ml, 30pcs/pk |
| YZSC1503-SAX | SAX, 150mg/3ml, 50pcs/pk | YZSC5006-SAX | SAX, 500mg/6ml, 30pcs/pk |
| YZSC10006-SAX | SAX, 1g/6ml, 30pcs/pk | | |

12. WCX

WCX is a weak cation exchange extraction column with silica as the matrix. The functional groups bonded to the silica surface are carboxyl groups, with a pKa of 3.8. Due to the presence of carboxylic acid groups, the anion exchange effect is not too strong, so it does not require highly alkaline eluents to elute the target compounds. WCX is particularly suitable for the adsorption and retention of strong cations, as strong cationic target substances interact significantly with SCX (Strong Cation Exchange) materials and are challenging to elute from SCXmaterials.

Technical parameters

| Structural Formula | Соон |
|---------------------|-----------------------------|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore |
| Function groups | Carboxylic acid group |
| Retention mechanism | Normal phase retention or |

e volume: 0.80 cm³/g Pore size: 60Å Surface area: 480m²/g oup r anion exchange

volume: 0.80 cm³/g Pore size: 60Å Surface area: 480m²/g

weak cation exchange

Detection and analysis of alkaloids and azacyclic compounds, such as paraquat in water, rapid detection of insecticides.

Ordering information of WCX

| P/N | Description | P/N | Description |
|--------------|---------------------------|---------------|--------------------------|
| YZSC1001-WCX | WCX, 100mg/1ml, 100pcs/pk | YZSC2003-WCX | WCX, 200mg/3ml, 50pcs/pk |
| YZSC603-WCX | WCX, 60mg/3ml, 50pcs/pk | YZSC5003-WCX | WCX, 500mg/3ml, 50pcs/pk |
| YZSC1003-WCX | WCX, 100mg/3ml, 50pcs/pk | YZSC5006-WCX | WCX, 500mg/6ml, 30pcs/pk |
| YZSC1503-WCX | WCX, 150mg/3ml, 50pcs/pk | YZSC10006-WCX | WCX, 1g/6ml, 30pcs/pk |

13. PRS

PRS is a strong cation exchange extraction column with silica as the matrix, bonded with propylsulfonic acid function- al groups. In non-polar solvents, PRS exhibits both polarity and hydrogen bonding interactions, making it suitable for the extraction and separation of cationic target substances.

Technical parameters

| Structural Formula | SO ³ H |
|---------------------|---|
| Matrix | Silica gel |
| Parameter | Particle size: 40-60µm Pore volume: 0.80 cm³/g Pore size: 60Å Surface area: 480m²/g |
| Function groups | Propane sulfonic acid group |
| Retention mechanism | Normal phase retention or weak cation exchange |

Application

Detection of cationic targets, alkaloids and azacyclic compounds.

Ordering information of PRS

| P/N | Description | P/N | Description |
|--------------|---------------------------|---------------|--------------------------|
| YZSC1001-PRS | PRS, 100mg/1ml, 100pcs/pk | YZSC2003-PRS | PRS, 200mg/3ml, 50pcs/pk |
| YZSC603-PRS | PRS, 60mg/3ml, 50pcs/pk | YZSC5003-PRS | PRS, 500mg/3ml, 50pcs/pk |
| YZSC1003-PRS | PRS, 100mg/3ml, 50pcs/pk | YZSC5006-PRS | PRS, 500mg/6ml, 30pcs/pk |
| YZSC1503-PRS | PRS, 150mg/3ml, 50pcs/pk | YZSC10006-PRS | PRS, 1g/6ml, 30pcs/pk |

Inorganic SPE

Inorganic SPE Adsorbents are commonly used as normal-phase polar adsorbents. Their polarities and surface acidities vary, which leads to different applications, but they are generally used for sample purification before analysis, especially for purifying organic extracts from complex samples. For example, Florisil adsorbents are frequently used to purify organic solvent extracts from plant and animal tissues, particularly in pesticide residue analysis. These adsorbents are also used for sample pretreatment in detecting Sudan dyes and malachite green in food products. Yicozoo Inorganic adsorbents are activated under strictly controlled conditions, ensuring effective purification with high and consistent recovery rates and excellent reproducibility.

1. Florisil

Florisil is a high-selectivity synthetic adsorbent composed of silica, magnesium oxide, and sodium sulfate. It is commonly used for sample purification and the extraction and separation of chlorinated pesticides before chromatographic analysis. Additionally, it is used for the extraction and separation of PCBs (polychlorinated biphenyls) and PAHs (polycyclic aromatic hydrocarbons).

Technical parameters

| Matrix | Magnesium silicate |
|---------------------|--------------------------|
| Parameter | 60-100 mesh (150-250 μm) |
| Function groups | Silicon hydroxyl |
| Retention mechanism | Normal phase retention |

Application

Analysis of environmental samples and pesticide residues, such as pyrethroid pesticide residues in eggs, detection of PAHs in water and food.

Ordering information of Florisil

| P/N | Description | P/N | Description |
|-------------------|--------------------------------|---------------------|-------------------------------|
| YZSC1001-Florisil | Florisil, 100mg/1ml, 100pcs/pk | YZSC5006-Florisil | Florisil, 500mg/6ml, 30pcs/pk |
| YZSC1003-Florisil | Florisil, 100mg/3ml, 50pcs/pk | YZSC10006-Florisil | Florisil, 1g/6ml, 30pcs/pk |
| YZSC1503-Florisil | Florisil, 150mg/3ml, 50pcs/pk | YZSC200012-Florisil | Florisil, 2g/12ml, 20pcs/pk |
| YZSC2003-Florisil | Florisil, 200mg/3ml, 50pcs/pk | YZSC500030-Florisil | Florisil, 5g/30ml, 10pcs/pk |
| YZSC5003-Florisil | Florisil, 500mg/3ml, 50pcs/pk | | |

2. Alumina-N

Alumina-N is a neutral alumina-based strong polar SPE adsorbent. The surface treatment renders it neutral, allowing it to interact with aluminum metal centers and form hydrogen bonds with surface silanol groups or perform ionic exchange with charged surfaces. It has a strong retention capacity for nitrogen, phosphorus, and sulfur-containing heterocyclic compounds, aromatic hydrocarbons, and organic amines. It is widely used for sample preparation in the analysis of Sudan dyes and malachite green.

Technical parameters

| Matrix | Al₂O₃ particle |
|---------------------|---------------------------|
| Parameter | Particle size: 50-200 μm |
| Function groups | Aluminum hydroxyl |
| рН | 7.5 |
| Retention mechanism | Normal phase retention or |

Application

Separation of polar or non-polar compounds from water-soluble and water-insoluble samples, or detection of food additives in beverages, such as lutein and acesulfame potassium, etc.

r anion exchange

Ordering information of Alumina-N

| P/N | Description | P/N | Description |
|---------------|---------------------------------|----------------|--------------------------------|
| YZSC1001-ALN | Alumina-N, 100mg/1ml, 100pcs/pk | YZSC2506-ALN | Alumina-N, 250mg/6ml, 30pcs/pk |
| YZSC1003-ALN | Alumina-N, 100mg/3ml, 50pcs/pk | YZSC5006-ALN | Alumina-N, 500mg/6ml, 30pcs/pk |
| YZSC1503-ALN | Alumina-N, 150mg/3ml, 50pcs/pk | YZSC10006-ALN | Alumina-N, 1g/6ml, 30pcs/pk |
| YZSC2003-ALN | Alumina-N, 200mg/3ml, 50pcs/pk | YZSC20006-ALN | Alumina-N, 2g/6ml, 30pcs/pk |
| YZSC3003-ALN | Alumina-N, 300mg/3ml, 50pcs/pk | YZSC100012-ALN | Alumina-N, 1g/12ml, 20pcs/pk |
| YZSC4003-ALN | Alumina-N, 400mg/3ml, 50pcs/pk | YZSC200012-ALN | Alumina-N, 2g/12ml, 20pcs/pk |
| YZSC5003-ALN | Alumina-N, 500mg/3ml, 50pcs/pk | YZSC400012-ALN | Alumina-N, 4g/12ml, 20pcs/pk |
| YZSC10003-ALN | Alumina-N, 1g/3ml, 50pcs/pk | YZSC500012-ALN | Alumina-N, 5g/12ml, 20pcs/pk |
| YZSC2006-ALN | Alumina-N, 200mg/6ml, 30pcs/pk | YZSC500030-ALN | Alumina-N, 5g/30ml, 10pcs/pk |

3. Alumina-B

Alumina-B is an alkaline alumina SPE adsorbent produced by treating alumina filler with an alkaline solution. The surface is negatively charged, providing functionality similar to cation exchange. The particle size of Alumina-B ranges from 50 to 200 µm

Technical parameters

| Matrix | Al ₂ O ₃ particle |
|---------------------|---|
| Parameter | Particle size: 50-200 μm |
| Function groups | Aluminum hydroxyl |
| рН | 10.0 |
| Retention mechanism | Normal phase retention or cation exchange |

Application

Retention of polar compounds, cationic compounds and neutral amine samples. Detection of sulfonamides in feed.

Ordering information of Alumina-B

| P/N | Description | P/N | Description |
|---------------|---------------------------------|----------------|--------------------------------|
| YZSC1001-ALB | Alumina-B, 100mg/1ml, 100pcs/pk | YZSC2006-ALB | Alumina-B, 1g/3ml, 50pcs/pk |
| YZSC1003-ALB | Alumina-B, 100mg/3ml, 50pcs/pk | YZSC5006-ALB | Alumina-B, 200mg/6ml, 30pcs/pk |
| YZSC1503-ALB | Alumina-B, 150mg/3ml, 50pcs/pk | YZSC10006-ALB | Alumina-B, 500mg/6ml, 30pcs/pk |
| YZSC2003-ALB | Alumina-B, 200mg/3ml, 50pcs/pk | YZSC100012-ALB | Alumina-B, 1g/6ml, 30pcs/pk |
| YZSC5003-ALB | Alumina-B, 500mg/3ml, 50pcs/pk | YZSC200012-ALB | Alumina-B, 1g/12ml, 20pcs/pk |
| YZSC10003-ALB | Alumina-B, 2g/12ml, 20pcs/pk | | |

4. Alumina-A

Alumina-A is an acidified alumina solid-phase extraction adsorbent with a surface pH of 4.5. The particle size ranges from 50 to 200 µm. It functions as a strong polar adsorbent and a moderate anion exchanger.

Application

Separation and purification of acid, moderate polarity and polar target compounds.

Ordering information of Alumina-A

| P/N | Description | P/N | Description |
|---------------|---------------------------------|---------------|--------------------------------|
| YZSC1001-ALA | Alumina-A, 100mg/1ml, 100pcs/pk | YZSC5003-ALA | Alumina-A, 500mg/3ml, 50pcs/pk |
| YZSC1003-ALA | Alumina-A, 100mg/3ml, 50pcs/pk | YZSC10003-ALA | Alumina-A, 1g/3ml, 50pcs/pk |
| YZSC1503-ALA | Alumina-A, 150mg/3ml, 50pcs/pk | YZSC2006-ALA | Alumina-A, 200mg/6ml, 30pcs/pk |
| YZSC2003-ALA | Alumina-A, 200mg/3ml, 50pcs/pk | YZSC5006-ALA | Alumina-A, 500mg/6ml, 30pcs/pk |
| YZSC10006-ALA | Alumina-A, 1g/6ml, 30pcs/pk | | |

5. Na₂SO₄

 Na_2SO_4 is a high-purity anhydrous sodium sulfate used as a drying agent to effectively remove moisture interference from samples. It offers superior cleanliness and dehydration performance compared to analytical-grade anhydrous sodium sulfate.

Ordering information of Na₂SO₄

| P/N | Description | P/N | Description |
|-------------------|-----------------------------|------------------|--|
| YZSC5003-Na2SO4 | Na₂SO₄, 500mg/3ml, 50pcs/pk | YZSC10006-Na2SO4 | Na ₂ SO ₄ , 1g/6ml, 30pcs/pk |
| YZSC600012-Na2SO4 | Na₂SO₄, 6g/12ml, 20pcs/pk | | |

6. Carb

Carb addresses the limitations of activated carbon, such as difficulty in eluting extracted substances, while retaining high affinity and large adsorption capacity for both polar and non-polar organic compounds. This results in excellent purification efficiency, high recovery rates, and high reproducibility. It is widely used in pesticide residue analysis and the prepro-cessing of samples with high pigment content. Additionally, unlike porous materials, graphitized carbon black achieves adsorp-tion equilibrium quickly, saving sample processing time.

Technical parameters

| Matrix | Graphite |
|---------------------|------------------------------|
| Function groups | Carbon six-member ring |
| Retention mechanism | Surface adsorption retention |

Application

Detection of moderate and nonpolar target compounds and aromatic ring compounds.

ion

Ordering information of Carb

| P/N | Description | P/N | Description |
|----------------|----------------------------|---------------|---------------------------|
| YZSC1001-Carb | Carb, 100mg/1ml, 100pcs/pk | YZSC2503-Carb | Carb, 250mg/3ml, 50pcs/pk |
| YZSC1003-Carb | Carb, 100mg/3ml, 50pcs/pk | YZSC5003-Carb | Carb, 500mg/3ml, 50pcs/pk |
| YZSC1503-Carb | Carb, 150mg/3ml, 50pcs/pk | YZSC2506-Carb | Carb, 250mg/6ml, 30pcs/pk |
| YZSC2003-Carb | Carb, 200mg/3ml, 50pcs/pk | YZSC5006-Carb | Carb, 500mg/6ml, 30pcs/pk |
| YZSC10006-Carb | Carb, 1g/6ml, 30pcs/pk | | |

7. Acticarbon

Acticarbon activated carbon column can be used for the detection of nitrosamine and acrylamide in water.

Ordering information of Acticarbon

| P/N | Description | P/N | Description |
|-------------|----------------------------------|---------------|---------------------------------|
| YZSC1001-AC | Acticarbon, 100mg/1ml, 100pcs/pk | YZSC2506-AC | Acticarbon, 250mg/6ml, 30pcs/pk |
| YZSC2003-AC | Acticarbon, 200mg/3ml, 50pcs/pk | YZSC5006-AC | Acticarbon, 500mg/6ml, 30pcs/pk |
| YZSC2503-AC | Acticarbon, 250mg/3ml, 50pcs/pk | YZSC100012-AC | Acticarbon, 1g/12ml, 20pcs/pk |
| YZSC1506-AC | Acticarbon, 150mg/6ml, 30pcs/pk | YZSC200012-AC | Acticarbon, 2g/12ml, 20pcs/pk |

Mixed Mode SPE

Mixed Mode SPE combine two types of stationary phases to utilize various interfacial effects for separating and purifying analytes. They are particularly useful for extracting basic drugs from biological matrices and analyzing pesticide residues in biological matrices, where numerous interfering substances are present and difficult to wash away. These mixed adsorbents are designed to handle complex sample matrices effectively.

1. C8/SCX

C8/SCX Mixed Mode SPE Product combines a C8 alkyl stationary phase and a strong cation exchange stationary phase SCX on a silica matrix in an optimized ratio. This dual-phase composition provides a combined retention mechanism. The C8 phase interacts with the hydrophobic parts of analytes, while the SCX phase interacts with the protonated amino groups of the analytes. Due to the strong dual interactions between the adsorbent and the analytes, it allows the use of stronger washing solvents and conditions to remove interfering substances adsorbed on the adsorbent.

Technical parameters

| Matrix | Silica gel |
|---------------------|--|
| Function groups | C8 alkyl chain, sulfonic acid group |
| Retention mechanism | Mixed mode of reversed-phase retention and cation exchange retention |

Application

Detection of cationic target compounds, such as melamine, clenbuterol, etc.

Ordering information of C8/SCX

| P/N | Description | P/N | Description |
|-----------------|------------------------------|------------------|-----------------------------|
| YZSC1001-C8/SCX | C8/SCX, 100mg/1ml, 100pcs/pk | YZSC5003-C8/SCX | C8/SCX, 500mg/3ml, 50pcs/pk |
| YZSC1003-C8/SCX | C8/SCX, 100mg/3ml, 50pcs/pk | YZSC2006-C8/SCX | C8/SCX, 200mg/6ml, 30pcs/pk |
| YZSC1503-C8/SCX | C8/SCX, 150mg/3ml, 50pcs/pk | YZSC5006-C8/SCX | C8/SCX, 500mg/6ml, 30pcs/pk |
| YZSC2003-C8/SCX | C8/SCX, 200mg/3ml, 50pcs/pk | YZSC10006-C8/SCX | C8/SCX, 1g/6ml, 30pcs/pk |

2. Carb/NH₂

Carb/NH₂ consists of an equal mix of graphitized carbon black and aminopropyl-bonded silica gel. This combi-nation provides unique separation and extraction capabilities, especially effective in pesticide residue analysis. It is particularly suited for removing pigments, fatty acids, and phenolic compounds, as well as extracting organophosphates from tea. This makes it ideal for preprocessing and analysis of pesticide residues in food and plant samples.

Technical parameters

| Matrix | Silica gel, graphitized carbo |
|---------------------|-------------------------------|
| Function groups | Carbon six-member ring, a |
| Retention mechanism | Mixed mode of reversed-p |

Application

Purification of samples in agricultural residue detection.

Ordering information of Carb/NH₂

| P/N | Description | P/N | Description |
|----------------|--|------------------|---|
| YZSC2/2506-CAN | Carb/NH ₂ , 250mg/250mg/6ml, 30pcs/pk | YZSC5/10006-CAN | Carb/NH ₂ , 500mg/1g/6ml, 30pcs/pk |
| YZSC5/5006-CAN | Carb/NH ₂ , 500mg/500mg/6ml, 30pcs/pk | YZSC1/100012-CAN | Carb/NH ₂ , 1g/1g/12ml, 20pcs/pk |
| YZSC3/5006-CAN | Carb/NH ₂ , 300mg/500mg/6ml, 30pcs/pk | | |

3. SAX/PSA

SAX/PSA is composed of two layers: the upper layer contains an equal amount of SAX (Strong Anion Exchange) material, and the lower layer contains an equal amount of PSA (Primary and Secondary Amine) material. The upper SAX layer adsorbs acidic substances from the sample matrix, while the lower PSA layer adsorbs organic acids, fatty acids, pigments, and other interfering substances. This dual-layer configuration is widely used for analyzing multiple pesticide residues in food.

Technical parameters

| Matrix | Silica gel |
|---------------------|-------------------------|
| Function groups | Quaternary ammonium gr |
| Retention mechanism | Mixed mode of normal ph |

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on

aminopropyl group

phase retention and cation exchange retention

| oup, ethylenediamine group |
|--|
| ase retention and anion exchange retention |

Purification of samples in agricultural residue testing detection.

Ordering information of SAX/PSA

| P/N | Description |
|---------------|------------------------------------|
| YZSC5/5006-SP | SAX/PSA, 500mg/500mg/6ml, 30pcs/pk |

4. Carb/PSA

Carb/PSA is a mixed-layer SPE product composed of an equal amount of graphitized carbon black in the upper layer and PSA (Primary and Secondary Amine) in the lower layer. The upper graphitized carbon black layer is effective at adsorb- ing pigments present in the sample matrix, while the lower PSA layer adsorbs organic acids, fatty acids, pigments, and other interfering substances. This dual-layer configuration is widely used for analyzing multiple pesticide residues infood.

Technical parameters

| Matrix | Graphitized carbon black |
|---------------------|---|
| Function groups | Carbon six-member ring, ethylenediamine group |
| Retention mechanism | Mixed mode of normal phase retention and surface adsorption |

Application

Purification of samples in agricultural residue detection.

Ordering information of SAX/PSA

| P/N | Description |
|---------------|-------------------------------------|
| YZSC5/5006-CP | Carb/PSA, 500mg/500mg/6ml, 30pcs/pk |

5. C18/CN

C18/CN columns are prepared by packing C18 and CN in specific proportions. They are suitable for the determination of four nitrofuran metabolites in seafood, including:1-Amino-2-oxazolidinone (AHD), 5-Methylmorpholine-3-amino-2-oxazolidi- none (AMOZ), 3-Amino-2-oxazolidinone (AOZ), Furazolidone metabolite amino-urea (SEM).

Application

Determination of nitrofuran metabolites residues in aquatic products by HPLC method.

Ordering information of C18/CN

| P/N | Description |
|-----------------|-----------------------------|
| YZSC2003-C18/CN | C18/CN, 200mg/3ml, 50pcs/pk |
| YZSC4003-C18/CN | C18/CN, 400mg/3ml, 50pcs/pk |
| YZSC6006-C18/CN | C18/CN, 600mg/6ml, 30pcs/pk |

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